

# Appendix

## Science 9 Two-Sided Data Sheet (Electricity Formulas on Back)

### Periodic Table of the First Eighteen Elements

1	1.01 1 <sup>+</sup> , 1 <sup>-</sup> <b>H</b> hydrogen	2	4.00 <b>He</b> helium
3	6.94 1 <sup>+</sup> <b>Li</b> lithium	4	9.01 2 <sup>+</sup> <b>Be</b> beryllium
5	10.81 3 <sup>+</sup> <b>B</b> boron	6	12.01 <b>C</b> carbon
7	14.01 3 <sup>-</sup> <b>N</b> nitrogen	8	16.00 2 <sup>-</sup> <b>O</b> oxygen
9	19.00 1 <sup>-</sup> <b>F</b> fluorine	10	20.18 <b>Ne</b> neon
11	22.99 1 <sup>+</sup> <b>Na</b> sodium	12	24.31 2 <sup>+</sup> <b>Mg</b> magnesium
13	26.98 3 <sup>+</sup> <b>Al</b> aluminum	14	28.09 <b>Si</b> silicon
15	30.97 3 <sup>-</sup> <b>P</b> phosphorus	16	32.07 2 <sup>-</sup> <b>S</b> sulfur
17	35.45 1 <sup>-</sup> <b>Cl</b> chlorine	18	39.95 <b>Ar</b> argon

  

**Legend for Elements**

Solid	Gas	Liquid
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**Note:** The legend denotes the states of elements at a temperature of 25 °C.

**Key**

3	6.94	1 <sup>+</sup>
Atomic number	molar mass	Common ion charges (most common first)
<b>Li</b>	lithium	
Symbol	Name	

  
